

4-(2-Chloro-4-nitrophenoxy)-*N*-[2-(4,6-dimethoxy-pyrimidin-2-yloxy)benzyl]anilineZheng-Bo Chen,<sup>a</sup> Jun Wu,<sup>a\*</sup>  
Pei-Zhi Zhang,<sup>b</sup> Pei-Min Zhang<sup>a</sup>  
and Jing Lv<sup>a</sup><sup>a</sup>Department of Chemistry, Zhejiang University, Hangzhou, Zhejiang, 310027, People's Republic of China, and <sup>b</sup>Department of Biological and Chemical Engineering, Zhejiang University of Science and Technology, Hangzhou, Zhejiang, 310012, People's Republic of China

Correspondence e-mail: wujunwjw@sohu.com

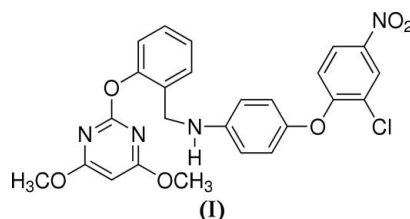
In the title molecule, C<sub>25</sub>H<sub>21</sub>ClN<sub>4</sub>O<sub>6</sub>, all bond lengths and angles show normal values. Weak intermolecular C—H···Cl hydrogen bonds link the molecules into chains extending along the crystallographic *c* axis. The crystal packing is further stabilized by van der Waals forces.

Received 17 March 2006  
Accepted 23 March 2006

## Comment

The derivatives of diphenyl ether show high herbicidal activities (Motoo *et al.*, 1995; Arnould *et al.*, 1998). The title compound, (I), has been prepared as a new herbicide. We report here its crystal structure.

## Key indicators

Single-crystal X-ray study  
*T* = 298 K  
Mean  $\sigma(\text{C—C}) = 0.003 \text{ \AA}$   
*R* factor = 0.048  
*wR* factor = 0.140  
Data-to-parameter ratio = 16.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In (I) (Fig. 1), the bond lengths and angles show normal values. The benzene C1–C6 (*A*), C7–C12 (*B*), C14–C19 (*C*) and pyrimidine (*D*) rings make the following dihedral angles: *A/B* = 80.93 (6)°, *B/C* = 82.73 (7)° and *C/D* = 85.70 (7)°.

Weak intermolecular C—H···Cl hydrogen bonds (Table 1) link the molecules into chains extending along the *c* axis. The crystal structure is further stabilized by van der Waals forces.

## Experimental

A mixture of 2-((4-(2-chloro-4-nitrophenoxy)phenylamino)methyl)phenol (0.20 g, 0.5 mmol), 2-methanesulfonyl-4,6-dimethoxy-pyrimidine (0.13 g, 0.5 mmol), and K<sub>2</sub>CO<sub>3</sub> (0.14 g, 1.0 mmol) in THF (40 ml) was refluxed for 8 h. After the insoluble substance had been removed from the mixture by filtration, the organic layer was evaporated *in vacuo* to give the crude product. Recrystallization of the crude product from ethanol gave yellow crystals (m.p. 391.3–392.2 K).

## Crystal data

C<sub>25</sub>H<sub>21</sub>ClN<sub>4</sub>O<sub>6</sub>  
*M<sub>r</sub>* = 508.91  
Triclinic, *P* $\bar{1}$   
*a* = 9.524 (4) Å  
*b* = 11.064 (4) Å  
*c* = 12.334 (7) Å  
 $\alpha$  = 88.869 (17)°  
 $\beta$  = 70.14 (2)°  
 $\gamma$  = 83.410 (19)°  
*V* = 1214.0 (10) Å<sup>3</sup>*Z* = 2  
*D<sub>x</sub>* = 1.392 Mg m<sup>-3</sup>  
Mo *K*α radiation  
Cell parameters from 8448 reflections  
 $\theta$  = 3.1–27.5°  
 $\mu$  = 0.21 mm<sup>-1</sup>  
*T* = 298 (1) K  
Chunk, yellow  
0.32 × 0.20 × 0.18 mm

Data collection

Rigaku RAXIS-RAPID  
diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan  
ABSCOR (Higashi, 1995)  
 $T_{\min} = 0.928$ ,  $T_{\max} = 0.964$   
11829 measured reflections

5464 independent reflections  
3282 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$   
 $\theta_{\text{max}} = 27.5^\circ$   
 $h = -10 \rightarrow 12$   
 $k = -14 \rightarrow 13$   
 $l = -15 \rightarrow 15$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.140$   
 $S = 1.05$   
5464 reflections  
327 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0735P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.22 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.30 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C13-H13A\cdots Cl1^i$	0.97	2.69	3.600 (3)	156

Symmetry code: (i)  $x, y, z + 1$ .

Atom H1, attached to N1, was located in a difference Fourier map and refined as riding in its as-found relative position, with a fixed  $U_{\text{iso}}$  of  $0.08 \text{ \AA}^2$ . The C-bound H atoms were placed in calculated positions, with C—H = 0.93  $\text{\AA}$  (aromatic), 0.96  $\text{\AA}$  (methyl group) and 0.97  $\text{\AA}$  (methylene group), and were refined as riding, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for aromatic and methylene H atoms or  $1.5U_{\text{eq}}(\text{C})$  for methyl.

Data collection: *PROCESS-AUTO* (Rigaku Corporation, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSM and Rigaku Corporation, 2004); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The project was supported by the National Natural Science Foundation of China (grant No. 20272052). We thank Jian-Min Gu for his assistance with the X-ray analysis.

References

Altomare, A., Casciarano, G., Giacovazzo, C. & Guagliardi, A. (1993). *J. Appl. Cryst.* **26**, 343–350.  
 Arnould, S., Takahashi, M. & Camadro, J.-M. (1998). *Biochemistry*, **37**, 12818–12828.  
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.  
 Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.  
 Motoo, S., Shinjiro, N., Harukazu, F., Takaharu, T., Ko, W. & Peter, B. (1995). *J. Agric. Food. Chem.* **43**, 1929–1934.

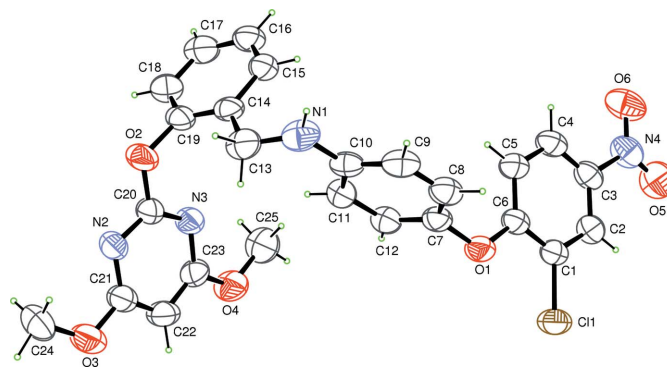


Figure 1  
The molecular structure of (I), with 40% probability displacement ellipsoids.

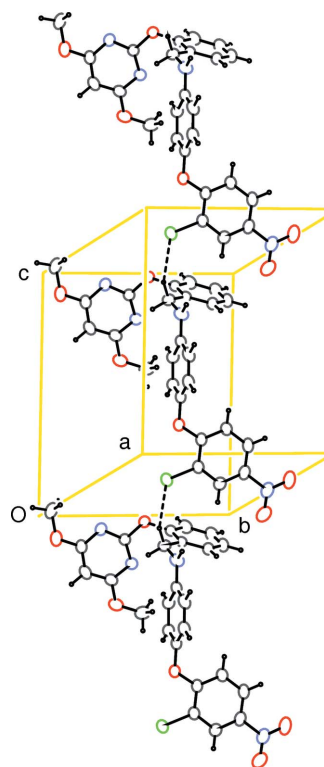


Figure 2  
The packing of (I), showing the weak intermolecular C—H...Cl hydrogen bonds as dashed lines.

Rigaku Corporation (1998). *PROCESS-AUTO*. Rigaku Corporation, Tokyo, Japan.  
 Rigaku/MSM and Rigaku Corporation (2004). *CrystalStructure* 3.6.0. Rigaku/MSM, The Woodlands, Texas, USA, and Rigaku, Tokyo, Japan.  
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.